

60/кислородом/Гибай

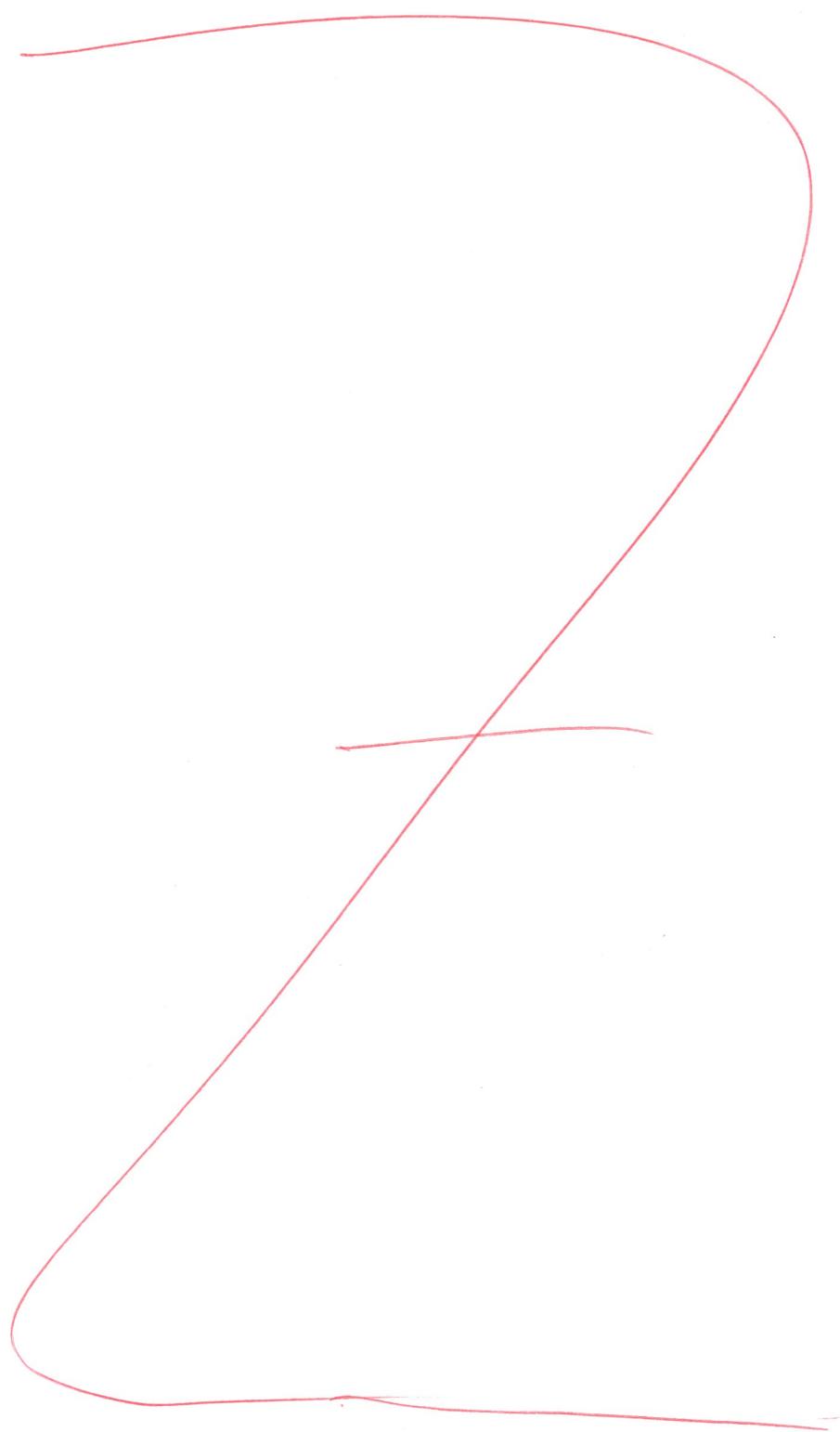
известок

Чистобик

1ug 6

N1

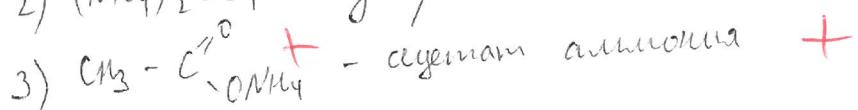
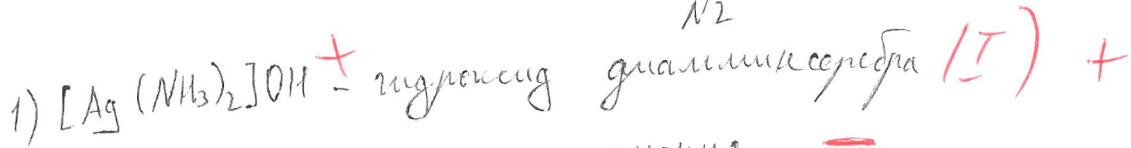
~~60~~ Mangan (As) + 65.



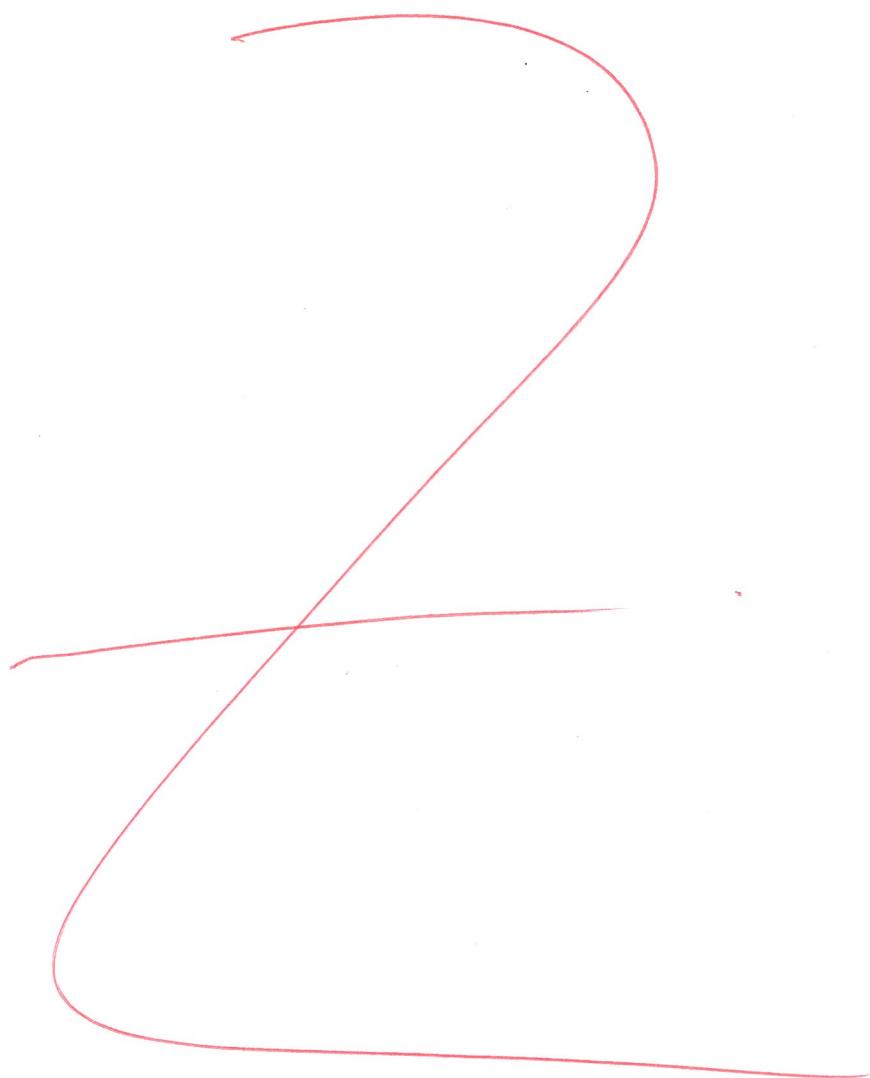
Чемодан

2 из 6

1/2



65.



Чемодан

N'3

X - N +

Y - S +

Z - Na +

$$0,069 = \frac{M(Z)}{M(Z)+310}$$

$$M(Z) = X$$

$$\frac{X}{X+310} = 0,069$$

$$\begin{aligned} X+310 &= 0,069 \\ X &= 18,39 \end{aligned}$$

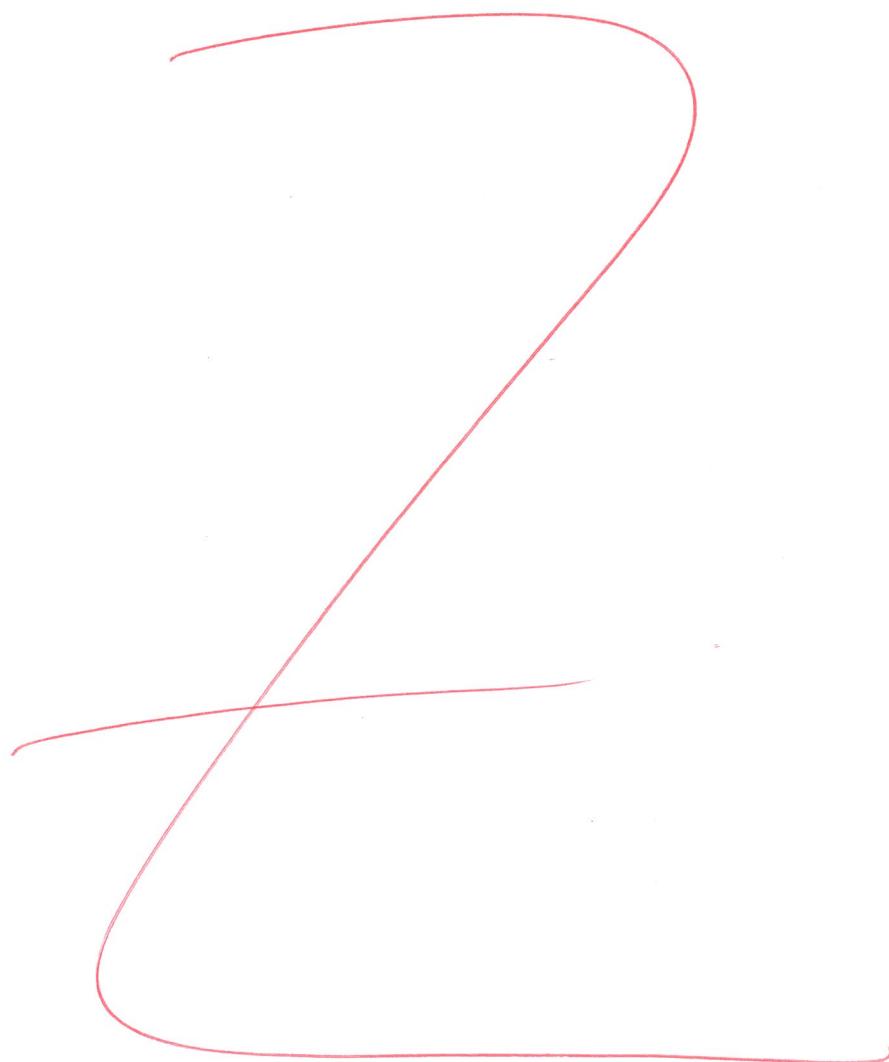
$$X - 0,069X - 18,39 = 0$$

$$0,931X = 18,39$$

$$X = 20,0$$

$$X - Na$$

185.



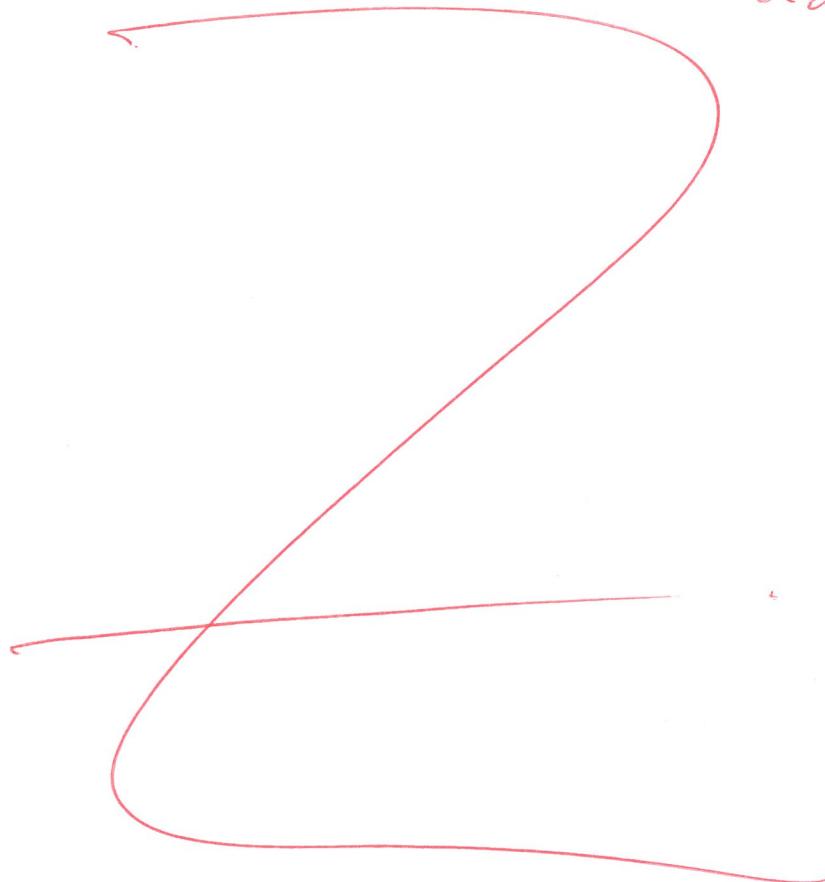
Чемодан

4 из 6

A - NH_3	+	X - HN_3^+	+
B - NO_2	+	Z - N_2O	+
C - HN_3	+	U - N_2H_4	+
D - HN_3	+	K - NF_3	+
E - N_2	+	W - NOF_3	+
		M - HCl	+
		H - $NaCN$	+

- N_4
- 1) $2NH_3 + \frac{5}{2}O_2 \xrightarrow{Pt} N_2O + 3H_2O$ $\xrightarrow{2NO + O_2} NO_2$ +
 - 2) $2NO_2 + H_2O \rightarrow HNO_3 + HNO_2$ +
 - 3) $HNO_2 + HN_3 \rightarrow N_2 + N_2O + H_2O$ +
 - 4) $12HNO_3 + P_4O_{10} \rightarrow 6N_2O_5 + 4H_3PO_4$ +
 - 5) $N_2O_5 + I_2 \rightarrow I_2O_5 + N_2$ +
 - 6) $2NH_3 + NaOCl \rightarrow N_2H_4 + NaCl + H_2O$ +
 - 7) $N_2H_4 + O_2 \rightarrow N_2 + 2H_2O$ +
 - 8) $2NH_3 + 3O_2 + 2CH_4 \rightarrow 2HCN + 6H_2O$ +
 - 9) ~~H_2O~~ $HCl + NaOH \rightarrow NaCl + H_2O$ +
 - 10) $NH_3 + 3F_2 \xrightarrow{600^\circ} 3HF + NF_3$ -
 - 11) $NF_3 + \frac{1}{2}O_2 \xrightarrow{-196^\circ} NOF_3$ +
 - 12) $NH_3 + HF \xrightarrow{?} NF_3 + ?$ -

238.



Часовник

5ug 6

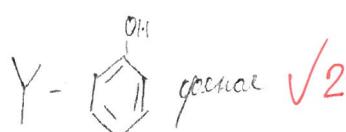
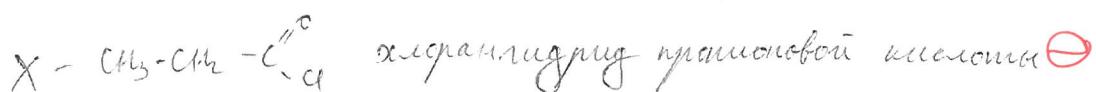
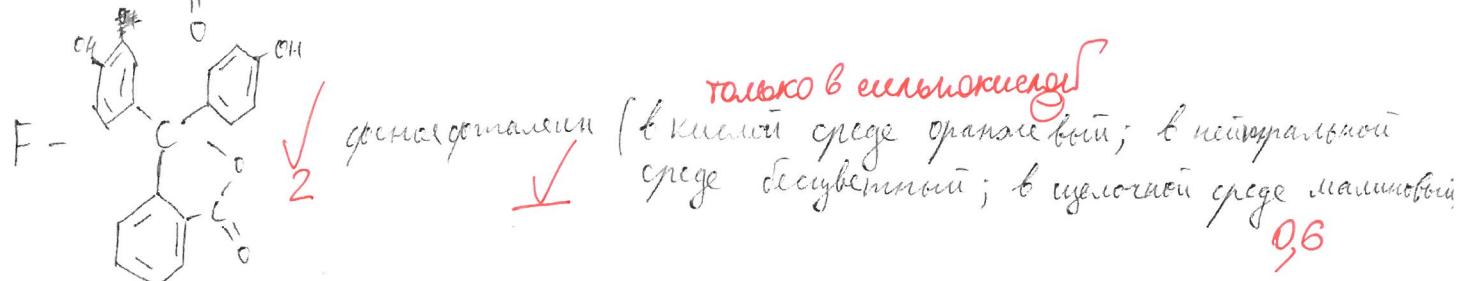
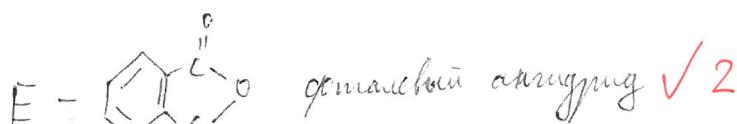
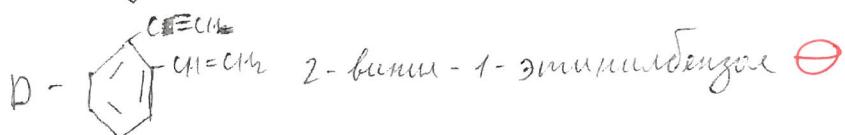
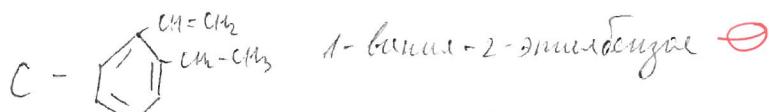
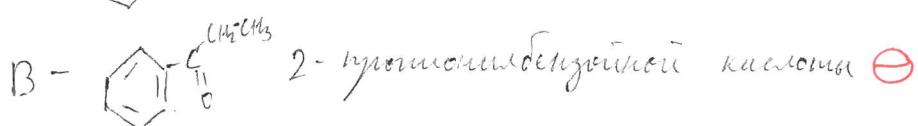
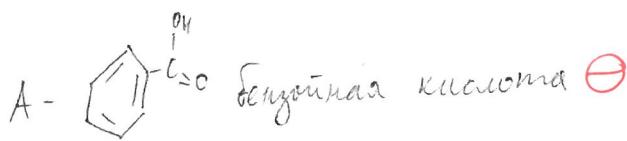
№5

- A - 2 минуты
короче?
- B -
- C -
- D -
- E -
- F -
- G -
- H -
- I -
- X -

(95)

Ученик
№ 6

Бугаев



G -

Σ660